

MATERIAL SAFETY DATA SHEET

PREMIER CHEMICALS

MSDS No.: 2623

Phone: PREMIER CHEMICALS: 1-800-227-4287
CHEMTREC, 24-Hr Emergency Assistance: 1-800-424-9300

Date Prepared: 3/05
This Revision:

SECTION 1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Material / Product Name(s): MAGOX® 93 WTG
CAS Number: 1309-48-4
Chemical Family: Mineral Oxide
General Use: A finely ground, chemical grade magnesium oxide powder.
Manufacturer / Supplier: PREMIER CHEMICALS
7251 Engle Road
Suite 415
Middleburg Heights, OH 44130

SECTION 2. INGREDIENTS / COMPOSITION

Table with 5 columns: Ingredient name, CAS Number, Percent, IARC/NTP/OSHA, Exposure Limits. Rows include Calcined Magnesite and Quartz\* (SiO2).

Typical Chemical Analysis, Wt.% (Loss Free Basis)

Table with 3 columns: Component, Wt.%, Range. Rows include LOI (1000°C), Acid Insol, R2O3, CaO, MgO.

The oxides shown in the typical chemical analysis do not exist in the magnesium oxide as free, uncombined oxides, but are combined mineralogically as calcium-magnesium silicates, aluminates and ferrites.

\*Quartz. Product may contain a trace of quartz, a polymorph of crystalline silica, which is classified by IARC as a "Known Human Carcinogen - Group 1".

SECTION 3. HAZARDS IDENTIFICATION

HMIS

Table with 2 columns: Hazard Category, Rating. Rows include Health Hazard, Flammability Hazard, Reactivity Hazard, Personal Protection.

EMERGENCY OVERVIEW:

A brownish, free flowing, fine granular material. Not a fire or spill hazard. Contact with water may cause product to swell, generate some heat, and burst its container.

Target Organs: Chronic overexposure may cause lung damage.
Primary route(s) of entry: Inhalation
Acute effects: Particulate may cause eye and upper respiratory irritation.

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### HAZARD IDENTIFICATION continued from page 1

**Chronic effects:** Product dust is classified as a "nuisance particulate, not otherwise regulated" as specified by ACGIH and OSHA. The excessive, long-term inhalation of mineral dusts may contribute to the development of industrial bronchitis, reduced breathing capacity, and may lead to the increased susceptibility to lung disease.

**Signs & symptoms of overexposure:**

**Eye contact:** Particulate is a physical eye irritant.

**Skin contact:** Low toxicity by skin contact.

**Inhalation:** Chronic overexposure by inhalation of airborne particulate may irritate upper respiratory system as well as the throat.

**Ingestion:** An unlikely route of exposure. If ingested in sufficient quantity, may cause gastrointestinal disturbances. Symptoms may include irritation, nausea, vomiting and diarrhea.

### SECTION 4. FIRST AID MEASURES

**Eye contact:** Flush eyes, including under the eyelids, with large amounts of water. If irritation persists, seek medical attention.

**Skin contact:** Wash affected areas with mild soap and water.

**Inhalation:** Remove victim to fresh air. If not breathing, give artificial respiration. Get immediate medical attention.

**Ingestion:** Ingestion is an unlikely route of exposure. If ingested in sufficient quantity and victim is conscious, give 1-2 glasses of water or milk. Never give anything by mouth to an unconscious person. Leave decision to induce vomiting to qualified medical personnel, since particles may be aspirated into the lungs. Seek immediate medical attention.

### SECTION 5. FIRE FIGHTING MEASURES

**NFPA code:** Flammability: 0, Health: 0, Reactivity: 0, Special: 0.

**Flash point:** Not Combustible

**Unusual Fire Hazard / Extinguishing Media:** Water reacts with magnesium oxide producing magnesium hydroxide and heat. Do not allow water to get inside containers; reaction with water will cause product to swell, generate heat, and burst its container. If contact is unavoidable, use sufficient water to safely absorb the heat that may be generated. Wetted product is not a health or environmental hazard.

**Hazardous Decomposition Products:** None

**Firefighting Instructions:** Firefighters should wear NIOSH-approved, positive pressure, self-contained breathing apparatus and full protective clothing when appropriate.

### SECTION 6. ACCIDENTAL RELEASE MEASURES

**Spill procedures:** Carefully, clean up and place material into a suitable container, being careful to avoid creating excessive dust from dried product. If conditions warrant, clean up personnel should wear approved respiratory protection, gloves, and goggles to prevent irritation from contact and/or inhalation.

### SECTION 7. HANDLING AND STORAGE

**Storage:** Store in dry, protected storage. Product is stable under normal conditions of dry storage. Do not allow water to get inside containers; reaction with water will cause product to swell, generate heat, and burst its container. Exposed, unprotected magnesium oxide will absorb moisture and carbon dioxide from the air. Minimize dust generation during material handling and transfer.

### SECTION 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

**Engineering controls:** Provide sufficient ventilation, in both volume and air flow patterns to control mist/dust concentrations below allowable exposure limits.

Page 2 --- EXPOSURE CONTROLS AND PERSONAL PROTECTION continues on page 3 --- Page 2

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### **EXPOSURE CONTROLS AND PERSONAL PROTECTION continued from page 2**

**Personal protective equipment:** The use of eye protection, gloves and long sleeve clothing is recommended.

**Respiration protection:** Provide workers with NIOSH approved respirators in accordance with requirements of 29 CFR 1910.134 for level of exposure incurred.

**Hygienic Practices:** Avoid contact with skin, eyes and clothing. After handling this product, wash hands before eating or drinking.

### **SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES**

**Appearance:** The product is brownish, free flowing powder; odorless.

**Boiling Point:** Not Applicable

**Specific Gravity (g/cc):** 3.56

**Melting Point:** >3800°F (>2100°C)

**Bulk Density (lbs./cu.ft.):** 60 - 70

**Water Solubility:** Slight <1%

**% Volatile by volume:** 0

**pH (10% aqueous slurry):** 10-11

**Evaporation rate:** Not Applicable

### **SECTION 10. STABILITY AND REACTIVITY**

**Hazardous Polymerization:** Will not occur

**Chemical Incompatibilities:** Magnesium oxide is soluble in aqueous acids generating heat and steam; violent reaction or ignition with interhalogens (e.g., bromine pentafluoride; chlorine trifluoride). Incandescent reaction with phosphorus pentachloride. Water will react with magnesium oxide producing magnesium hydroxide and heat.

**Hazardous Decomposition Products:** Heat and Steam

### **SECTION 11. TOXICOLOGICAL INFORMATION**

**Magnesium Oxide** CAS #1309-48-4 Toxic and Hazard Review: low toxicity - a nutrient and/or dietary supplement food additive. THERAP CAT: antacid. (Sax) an experimental tumorigen. Inhalation of fume (not MgO dust particular) produced upon decomposition of magnesium compounds can produce a febrile reaction and leukocytosis in humans.

TOXICITY DATA: ihl-hmn TCLo:400mg/m<sup>3</sup>; itr-ham TDLo:480 mg/kg/30w-I:ETA.

**Quartz** CAS #14808-60-7. Toxic and Hazard Review (Sax): Experimental poison by intratracheal and intravenous routes. An experimental carcinogen, tumorigen, and neoplastigen. Human systemic effects by inhalation: cough, dyspnea, liver effects. Listed by IARC as a "Known Human Carcinogen" Group 1. Listed by NTP.

TOXICITY DATA: No LD<sub>50</sub> in RTECS. ihl-hmn: TCLo 16 mppcf / 8 hrs / 17.9Y-I: PUL; ihh:hmh LCLo: 300 µg / m<sup>3</sup> / 10 Y-I;LVR; Other species toxicity data (NIOSH RTECS): inv-rat LDLo: 90mg/kg; itr-rat LDLo: 20mg/kg; inv-mus LDLo: 40mg/kg; inv-mus: 20mg/kg.

### **SECTION 12. ECOLOGICAL INFORMATION**

**Ecotoxicological / Chemical Fate Information:**

No data available on any adverse effects of this material on the environment.

### **SECTION 13. DISPOSAL INFORMATION**

**Waste Management/Disposal:** This product does not exhibit any characteristics of a hazardous waste. The product is suitable for landfill disposal. Follow all applicable federal, state and local regulations for safe disposal.

### **SECTION 14. TRANSPORT INFORMATION**

**US Department of Transportation:** Not regulated by DOT as a hazardous material. No hazard class, no label or placard required, no UN or NA number assigned.

**Canadian TDG Hazard Class & PIN:** Not regulated.

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### SECTION 15. REGULATORY INFORMATION

**SARA TITLE III:** This product does not contain any substances reportable under Sections 302, 304 or 313. Sections 311 and 312 do apply. (Routine Reporting and Chemical Inventories)

**TSCA:** All substances in this product are listed in the Chemical Substance Inventory of the Toxic Substances Control Act.

**CERCLA Hazardous Substance List, RQ:**No

**California Proposition 65:** This product contains chemicals known to the State of California to cause cancer, birth defects or other reproductive toxins.

### SECTION 16. OTHER INFORMATION

#### ACRONYMS AND REFERENCES USED IN PREPARATION OF MSDS':

ACGIH:	American Conference of Governmental Industrial Hygienists
CAS#:	CAS Registration Number is an assigned number to identify a material. CAS stands for Chemical Abstracts Service.
CERCLA:	Comprehensive Environmental Response, Compensation & Liability Act
EPCRA:	Emergency Planning and Community Right-to-Know Act of 1986
HMISTM:	Hazardous Materials Identification System (National Paint & Coatings Association)
IARC:	International Agency for Research on Cancer
MSHA:	Mine Safety and Health Administration
mg/m <sup>3</sup> :	Milligrams per cubic meter
NIOSH:	National Institute for Occupational Safety and Health
NFPA:	National Fire Protection Association
NTP:	National Toxicology Program
OSHA:	Occupational Safety and Health Administration
PEL:	Permissible Exposure Limit (OSHA)
REL:	Recommended Exposure Limit (OSHA)
SARA:	Superfund Amendments and Reauthorization Act
TITLE III:	Emergency Planning and Community Right-to-Know Act
Section 302:	Extremely Hazardous Substances
Section 304:	Emergency Release
Section 311:	<i>Community Right-to-Know</i> , MSDSs or List of Chemicals
Section 312:	<i>Community Right-to-Know</i> , Inventory and Location, (Tier I/II)
Section 313:	Toxic Chemicals, Toxic Chemical Release Reporting, Form R
TLV:	Threshold Limit Values (ACGIH)
TWA:	Time Weighted Average
29CFR1910.134:	OSHA Respiratory Protection Standard

#### REFERENCES:

Sax, N. Irving: Dangerous Properties of Industrial Materials, Ninth Edition, Van Nostrand Reinhold Co., Inc., 1996.  
Kirk, R. and Othmer, D., Encyclopedia of Chemical Technology, Third Edition, Wiley-Interscience, New York, NY 1982.  
Clansky, K.B., Suspect Chemicals Sourcebook, 1992-2<sup>nd</sup> Edition, Roytech Publications, Bethesda, Maryland.  
Sax, N. Irving and Lewis, R.J. Hawley's Condensed Chemical Dictionary, Eleventh Ed., Van Nostrand Reinhold Co., Inc., NY  
Manufacturers / Suppliers, Material Safety Data Sheets on Raw Materials Used  
American National Standard for Hazardous Industrial Chemicals - Material Safety Data Sheets - Preparation, American National Standards Institute, Inc., 11 West 42<sup>nd</sup> St, New York, NY 10036.

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